Class: 6 Chemistry: Chapter-2 The Nature of Matter(States of Matter) Answersheet

- I.Name the following:
- 1. Atom 2. Molecule. 3. Matter. 4. Paper 5. Wood.
- II. Fill in the blanks:
- 1.Mass 2. Definite. 3. Solids. 4. Solids. 5. Gas
- III. True or false:
- 1. False. 2. True. 3. True. 4. True. 5. False
- IV Complete the following exercises given in the book:

Page 33 worksheet:

- Q I. Fill in the blanks:
- 1. A Molecule. 2. Gases. 3. Gas 4. Volume, shape 5. Compressed
- QII. Match the following:
- 1-c. 2-e. 3-a. 4-b. 5-d
- III. True or false
- 1. True. 2. True 3. False 4. False. 5. True

Page 35,36 and 37:

- A. Tick the correct answer:
- 1-a. 2-c. 3-c. 4-a. 5-c. 6-a. 7-a. 8-a
- B. Fill in the blanks:
- 1. Solid liquid and gas. 2. Lose. 3. Solids. 4. Fixed. 5. Gas. 6. Mass, space
- 7. Gases
- C. True or false. Correct the false statements:
- 1. True
- 2. True.
- 3. False. Nylon and steel are man made materials.
- 4. True.
- 5.True.
- 6. False. John Dalton discovered that a molecule is made up of smaller particles called atoms.
- D. Find the odd one out giving reason:
 - 1. Water. It is a liquid, while the others are gases.

- 2. Protons. These are sub-atomic particles where as the others are three States of matter.
- 3. Atom. it is the smallest particle of matter where as the others are sub- atomic particles.
- 4. Naphthalene. It is a sublimating substance while the others are three different states of water.
- 5. Sublimation. Because in this process solid directly changes into gaseous state skipping the liquid state.

E. Match the columns:

1-f. 2-d. 3-e. 4-b. 5-c. 6-a

- F. Give reasons:
- 1. Gases have low densities. (Refer to conceptual Questions: ans 4)
- 2. Heat energy bring change of state.(Refer to conceptual Questions: ans 6)
- 3. Solids do not flow.(Refer to Conceptual Questions: ans 2)
- 4. Liquids can be compressed.(Refer to Conceptual Questions: ans 3)
- G. Answer the following questions:
- 1. Matter is anything which occupies space, has weight and can be perceived by our senses.
- 2. Solids cannot be compressed because the molecules are closely packed and they have least intermolecular spaces.
- 3. Page 28 difference in tabular form
- 4. Yes. Atom is divisible into electrons, protons and neutrons.
- 5. Page 30 Activity 4
- 6. The force of attraction between the molecules is very strong in solids, lesser in liquids and least in gases.
- 7. Page 30 Activity 5
- 8. Page 29 The kinetic theory of matter: Definition and all the points.
